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## Enhanced Validation of Advanced Battery Supply Chains (EVALS) Overview

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Photo by Josh Bauer, NREL 61725



Mining and refining can take years to develop because mines have different impurities (unless supplier can leverage existing mine). Materials producers require years to develop battery grade precursors that optimize processing for purity and performance. Once material suppliers meet qualifications, many more years are required for validation in specific applications.

## EVALS: Accelerate the path from mine to battery and improve supply chain resiliency



#### EVALS primary goal is to reduce the time and cost associated with adopting domestic sources by:

Better understanding source impurity impacts on material properties and performance to inform qualification decisions and cost-performance tradeoffs.

Developing suite of tools and processes to decrease validation time for domestic battery materials to support material manufacturers and automative EV OEMs.

## LiFePO<sub>4</sub> (LFP) was selected as an initial example material for EVALS, but EVALS is *not* an LFP project.

#### US gigafactories online FY26 show interest in US manufacturing (all chemistries).<sup>1</sup>

AA Portable Power AKASOL Alion Science and Technology AllCell Technology Altair Nanotechnologies American Battery Factory American Battery Solutions American Lithium Energy Auto Motive Power **Battery Specialties** Blue Line Battery **BMW Spartanburg Plant** BorgWarner-Romeo Power **Braille Battery** Cell-Con Clarios Conamix **Coreshell Technologies** Cuberg

Custom Power Dantona Industries EaglePicher Technologies **Electric Power Systems** Electrochem Solutions EnerSys EnerSys **Enovate Medical** eNow EnPower Envia Systems **Envision AESC Smyrna Plant Exponential Power Exponential Power Exponential Power** Factorial Flexodes Flux Power

Ford Rawsonville Plant Ford-SK BlueOval Ford-SK BlueOval Citv General Motors Brownstown Battery GM-LG Ultium Cells **GM-LG Ultium Cells GM-LG Ultium Cells GS** Yuasa Lithium Power **Highpower International** Honda Brownstone Battery Hyundai-SK Envision AESC Bowling Green Plant Inventus Power Woodridge Facility Kore Power LG Energy LG Energy Lithion Battery Lithion Battery Lithion Battery LytEn

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1. https://electronsx.com/battery-gigafactories.php

LFP demand is growing rapidly and US demand projected >4x higher in 2030 (180 GWh/y) than 2025.<sup>2</sup> LFP contains fewer critical materials than conventional  $LiNi_{x}MnyCo_{1-x-y}O_{2}$  (NMC) – more available in US. NREL | 4

2. Source: BloombergNEF https://cen.acs.org/energy/energy-storage-/Lithium-iron-phosphate-comes-to-America/101/i4









## Iron resource case study: Minnesota Mesabi Iron Range mined since late 19<sup>th</sup> century









By W.F. Cannon (USGS) Public Domain, https://commons. wikimedia.org/w/index.php?curid=39161950

By Chipcity - Own work, CC BY-SA 3.0, https://commons.wikimedia.org/w/index.php?curid=21362665 All Fe currently mined to make low-silica magnetite pellet product for steel manufacture.

Taconite is the major orebody

Infrastructure, workforce, and

mineral mined in the region.

transportation networks.

Fe required for LFP is <<1% of that required for steel.

#### Mining, Comminution, and Beneficiation Flowsheet to Form Iron Ore Product for Steelmaking

 $Drill \rightarrow Blast \rightarrow Shovel \rightarrow Haul \rightarrow 1^{\circ} \text{ and } 2^{\circ} \text{ Gyratory Crushing} \rightarrow Rail \text{ Transport to Concentrator} \rightarrow 3^{\circ} \text{ and } 4^{\circ} \text{ Cone Crushing} = 10^{\circ} \text{ Gyratory Crushing} \rightarrow Rail \text{ Transport to Concentrator} \rightarrow 3^{\circ} \text{ and } 4^{\circ} \text{ Cone Crushing} \rightarrow 10^{\circ} \text{ Gyratory Crushing} \rightarrow 10^{\circ} \text{ Gyr$ 

Rod Milling  $\rightarrow$  Dry Magnetic Separation  $\rightarrow$  Ball Milling  $\rightarrow$  Rougher Magnetic Separation  $\rightarrow$  Gravity Separation  $\rightarrow$  Screening

 $\rightarrow$  H<sub>2</sub>O Recovery  $\rightarrow$  Finisher Magnetic Separation  $\rightarrow$  Thickening  $\rightarrow$  Filtration  $\rightarrow$  Balling  $\rightarrow$  Binding  $\rightarrow$  Pelletization

Counterproductive in sourcing Fe for LFP

### Iron resource case study: orebody and processing impurities

Process Knowledge and Sourcing

Apply knowledge of **geology** and **processing** to domestic primary resources to understand feasibility for **battery precursor** products.

#### Iron Ore Pellet Composition Targets for Steelmaking

Composition (weight %)	Min	Target	Max	
Fe	65			
Silica	4.60	4.80	5.00	
CaO	0.77	0.85	1.20	
Р	0.016	0.021	0.021	
$Na_2O + K_2O$		0.062	0.073	
Moisture (H <sub>2</sub> O)		2.75	4.30	



Sadeghi, B., Najafizadeh, M., Cavaliere, P., Shabani, A. and Aminaei, M., 2024. Effect of composition and processing conditions on the direct reduction of iron oxide pellets. Powder Technology, 444, p.120061. (open access)

# LFP has different impurity concerns than steel.

Magnetite concentrate products are analyzed for suitability for steel production, not for alternative minor products. Example: P induces steel embrittlement, so it must be minimized in steel, but small P impurity may not degrade LFP.

Also have to consider impurity impacts from processing agents (clay binders, frothing agents, flocculants, etc.).

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## When converting Fe ore to LFP precursors, what impurities are likely?





Taconite iron ore pellets for compositional and leaching studies.

Most common Fe ore impurities are refractory (high melting temperatures and chemically resistant).

Silica, alumina, magnesia not mobilized in leaching.

Aqueous matrix matters for impurity removal (e.g., Ca).
Sparingly soluble in sulfate environment as CaSO<sub>4</sub>.
Moderately soluble in basic environment as Ca(OH)<sub>2</sub>.
Highly soluble in chloride environment as CaCl<sub>2</sub>.

Solvent extraction oxidizes Fe to Fe<sup>3+</sup> and other trivalent ions tend to accompany (e.g., Cr<sup>3+</sup>, Al<sup>3+</sup>).

Near neighbor elements are often difficult to separate.





Accelerating material sources from mine to batteries requires an example LFP synthesis process to introduce impurities



Impurity identification from source analysis



Downselect and send materials for electrode and cell fabrication

#### 98.5% of LFP is made in China today by a two-step process.





 $FeSO_4 + H_3PO_4 \rightarrow FePO_4$ 

 $FePO_4 + Li_2CO_3 + dextrose \rightarrow LiFePO_4-C$ 

 $FeSO_4$  is waste biproduct in  $TiO_2$  manufacturing in China<sup>1-3</sup> that minimizes cost and uses local resources.

#### Synthesis approaches and considerations vary in USA.

Precursor cost/availability and environmental regulations in US differ from China.

Some targeting Fe/Fe oxides to avoid sulfate waste<sup>1-2</sup> or purchasing Chinese-manufactured FePO<sub>4</sub>.

Impurity impacts depend on synthesis process.

## EVALS using two-step process to demonstrate tools for validating sources and accelerating mine to battery path.

1. https://cen.acs.org/energy/energy-storage-/Lithium-iron-phosphate-comes-to-America/101/i4 2. Y. Caprara, "Identifying the pinch points in the LFP supply chain," Battery Materials Review (2022), 5 (2). 3. J. Quan, S. Zhao, D. Song, T. Wang, W. He, G. Li, Sci. of Total Environ. 819 (2022) 153105. NREL

# Correlating materials characterization and electrochemical performance may inform rapid screening protocols





Train models with materials and electrochemical characterization.

# Correlating materials characterization and electrochemical performance may inform rapid screening protocols





Train models with materials and electrochemical characterization.

Reduce electrochemical matrix with materials characterization screening tools and protocols.

# Correlating materials characterization and electrochemical performance may inform rapid screening protocols





Train models with materials and electrochemical characterization.

Reduce electrochemical matrix with materials characterization screening tools and protocols.

Need to link materials signatures to performance.

Four **commercial LFP** materials show vary different performance in low loading half cells.

Characterizing to correlate material properties with performance and most effective screening tools.



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## Commercial LFPs show distinct bulk physical and chemical signatures

Vendor 1

Vendor 2



Performance not simply correlated to particle size, surface area, or tap density (despite large variation).

Poor performance correlated with higher than 1:1 Fe:P and presence of S impurities.





Vendor 3



**Chemical Characterization** 



### Solid state NMR sensitive to atomic and electronic <sup>6</sup>Li and <sup>31</sup>P environments and EPR sensitive to magnetic and electronic structure of Fe



EPR and NMR sensitive to changes between commercial LFPs but still determining relation to structural features.

NMR = nuclear magnetic resonance EPR = electron paramagnetic resonance



Sideband manifold span informs local symmetry and interactions between paramagnetic metals.





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### Probing Performance Differences with STEM/EDS and 4D STEM





Particle-scale compositional heterogeneity evident.

C/Ti coating uniformity varies.

Atomic-level defects (vacancies, anti-site disorder, substitutions).

STEM = scanning transmission electron microscopy EDS = energy dispersive spectroscopy







Vendor 3

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### High quality electrode coatings and cells from CAMP will enable robust electrochemical testing and statistically relevant results

**Cell and Electrode** Development

LFP slurry optimization

Coating development

#### Cell design and fabrication

Roll-to-roll doctor blade with slurry hopper not ideal for high surface area materials but only requires 20-50 g of material.

Commercial and **EVALS-synthesized** LFP materials.



CAMP = Argonne's Cell Analysis, Modeling, and **Prototyping Facility** 



Actively pumps slurry through slot so ideal for high surface area electrodes but need hundreds of g of material.



CAMP capable of semi-automated pilot-scale manufacturing in dry room.





92% active LFP, 2.3 mAh/cm<sup>2</sup>, 35% porosity, N/P = 1.05-1.1









## Tiered testing protocols help reduce long-term cycling test matrix



Support rapid decision-making on impurities using approaches that combine both electrochemical and non-electrochemical data. Innovate testing diagnostics to accelerate testing while preserving key signatures that enable machine learning/artificial intelligence approaches.





## Develop models to predict lifetime, reduce test matrix/time, and inform costperformance tradeoffs

### **Traditional life testing**

Batch testing cells in test matrix by cycling replicate cells to failure in parallel is inefficient use of time and channels.



Channel	Testing Weeks								
1									
2									
3									

### DoE and early life prediction

Models and Predictions

Develop early prediction models and employ them to reduce test matrix and inform next experiment.





#### **Cost-performance tradeoffs**

Inform TEA to understand cost impacts of bad-actor impurity removal.



Couple material purity and processing information with performance.



TEA = technoeconomic analysis



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## Questions? Comments? Thank you.

#### www.nrel.gov

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## Why focus on impurities and tool development?

#### **Identified Challenges**

- Each geologic source has different impurity levels.
- New source validation is time consuming.
- Low uniformity in purification methods and tolerance levels across materials production and cell manufacturing.
- High-throughput, combinatorial analysis is insufficient and costly.



### **Opportunities for Tools to Save Time and Cost**

- Location-specific forecasting of impurities and impacts.
- Test matrix reduction methods decrease time and resources for combinatorial, high-throughput approaches.
- Coordinated characterization, testing, prediction, and modeling tools.
- Expanded parameter extraction from early data.
- Integrated tools strengthen links between performance and cost as a function of processing and impurities.
- Transferable insights across materials and cell designs.
- Potential for qualification standardization.



Process Knowledge and Sourcing



Aldrich, C. and Liu, X., 2021. Monitoring of flotation systems by use of multivariate froth image analysis. Minerals, 11(7), p.683. (open access)

#### Iron Ore Pellet Production: Entrained Reagents and Consumables

Consumable	lb/Mt pellet product			
Grinding Media (Fe)	4.48			
Diamine $(CH_2(NH_2)_2)$	0.144			
Flocculant $[(C_6H_{11}XO_4)_n]$	0.027			
Soda Ash (Na <sub>2</sub> CO <sub>3</sub> )	1.557			
Frother (CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH(CH <sub>2</sub> CH <sub>3</sub> )CH <sub>2</sub> OH)	0.027			
Bentonite [(Na,Ca) <sub>0.33</sub> (Al,Mg) <sub>2</sub> (Si <sub>4</sub> O <sub>10</sub> )(OH) <sub>2</sub> ·nH2O]	9.0			
Organic ( $C_x H_y O_z$ )	0.50			
Fluxstone (Ca-rich melting-point depressant)	26.93			

Compositional targets ignore the reagents used in separations.

Entrained reagents/impurities problematic in downstream *steel production* are tracked; these will differ for battery precursors.